

Sub- T_g glass relaxation, nucleation, and crystallization in lithium disilicate glass

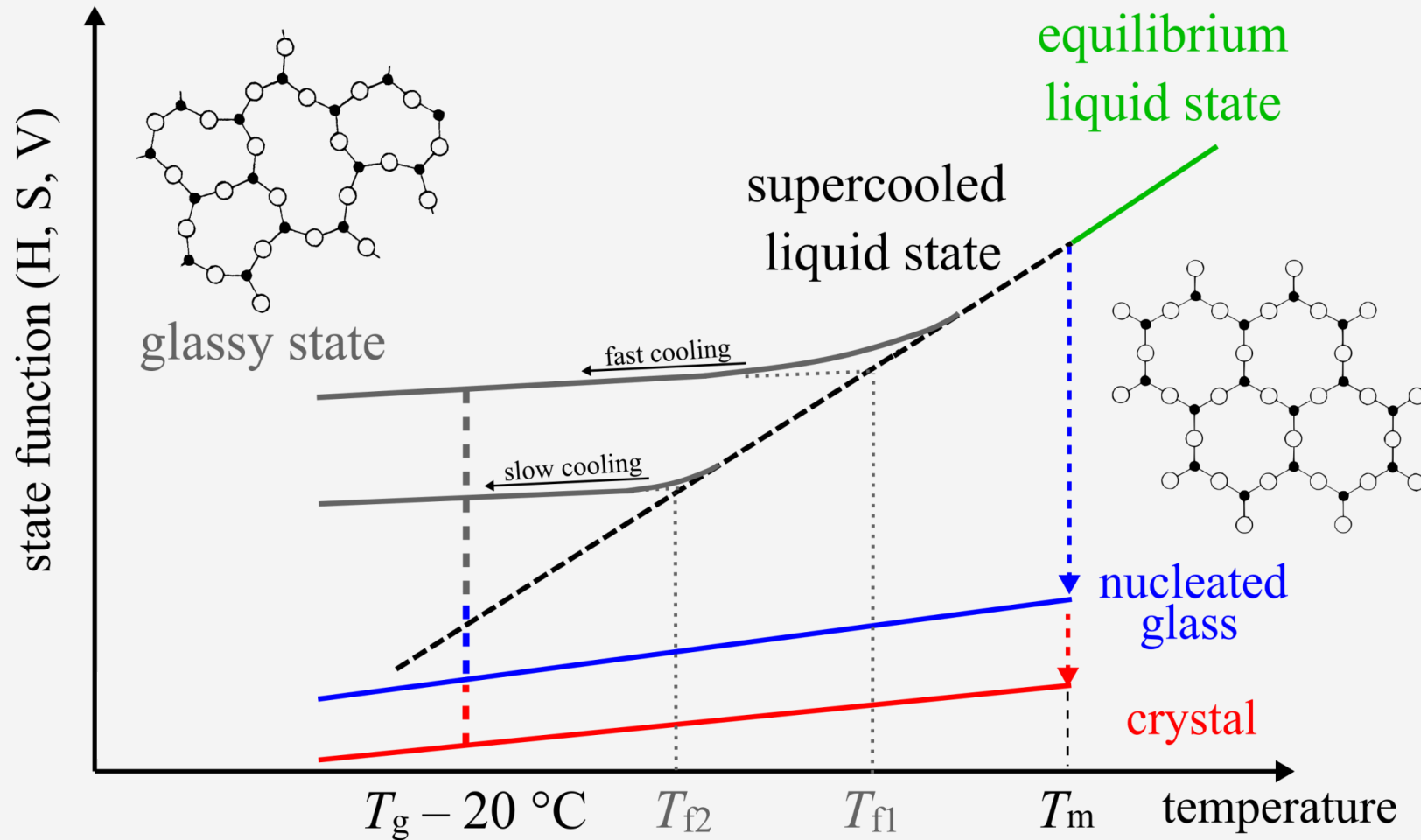
A solid-state NMR and MD strategy

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What happens structurally when glasses head towards equilibrium?



Experimental Relaxation and Nucleation Study: $\text{Li}_2\text{Si}_2\text{O}_5$ glass

1.

Glass synthesis

Melting at 1400 °C for 3h
Splat cooling @ RT
Repeated three times
Colorless, homogeneous
glass

2.

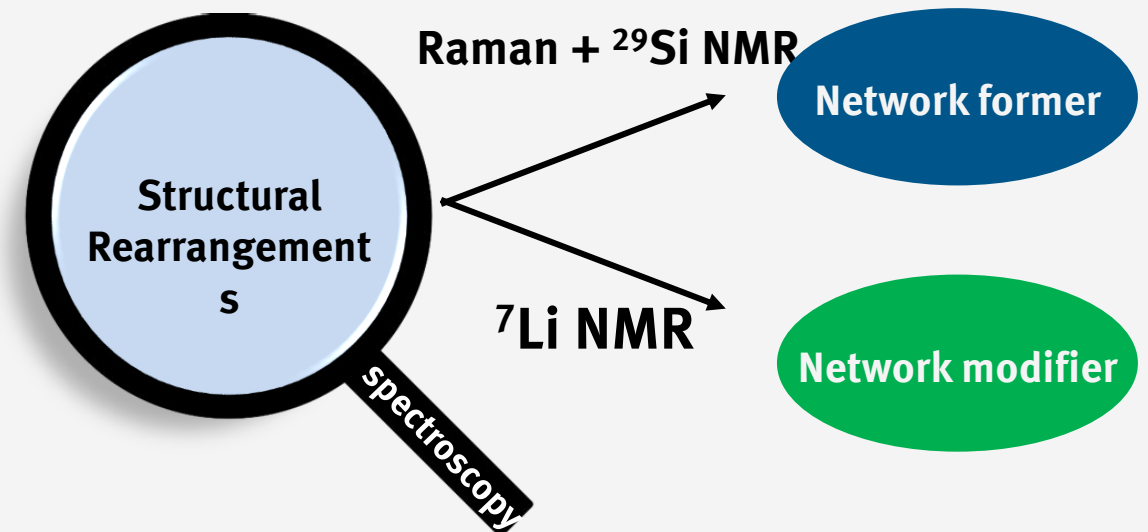
Annealing protocol

2 g pieces, vertical furnace
435 °C (ca. $T_g - 20^\circ\text{C}$)
15 min to 60 d
Cooled quickly (ambient
cond.)

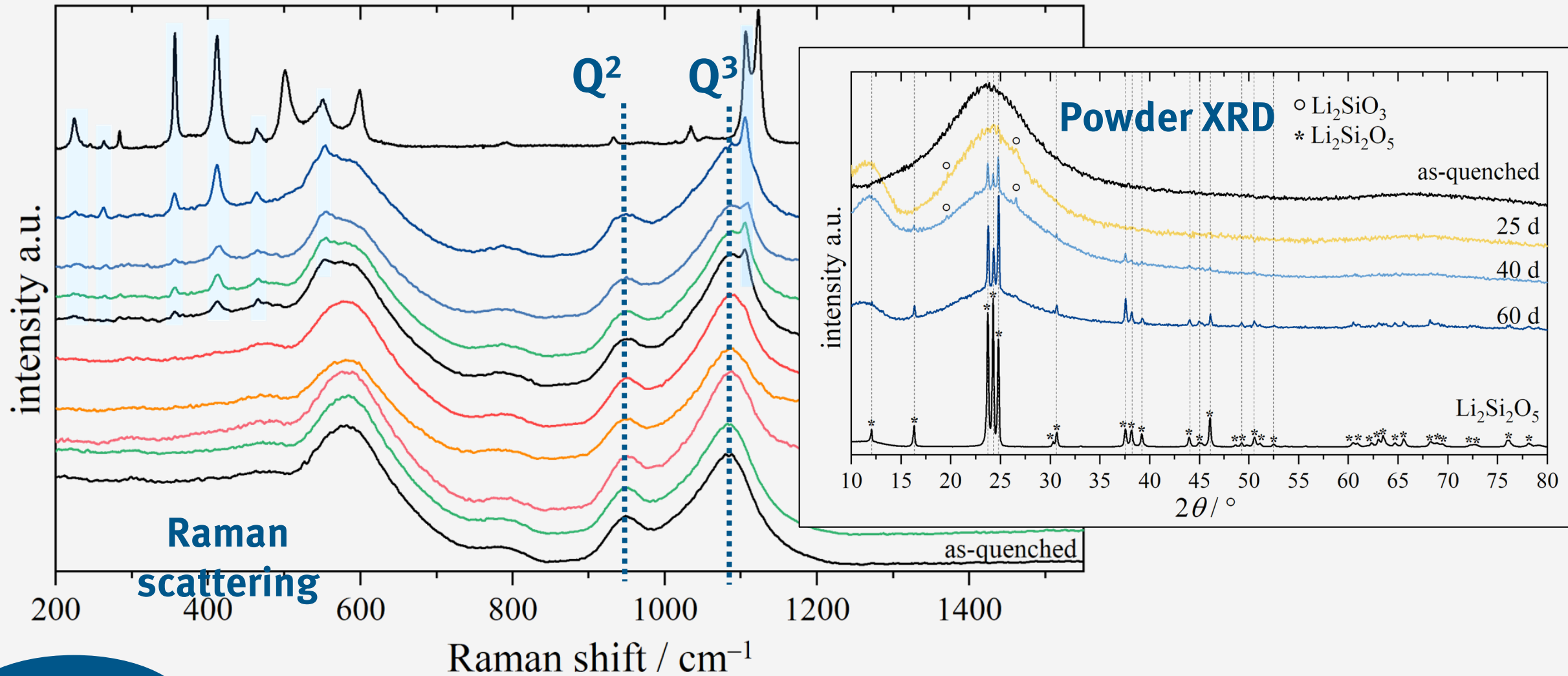
3.

Characterization

DSC, XRD
Raman
NMR
MD simulations

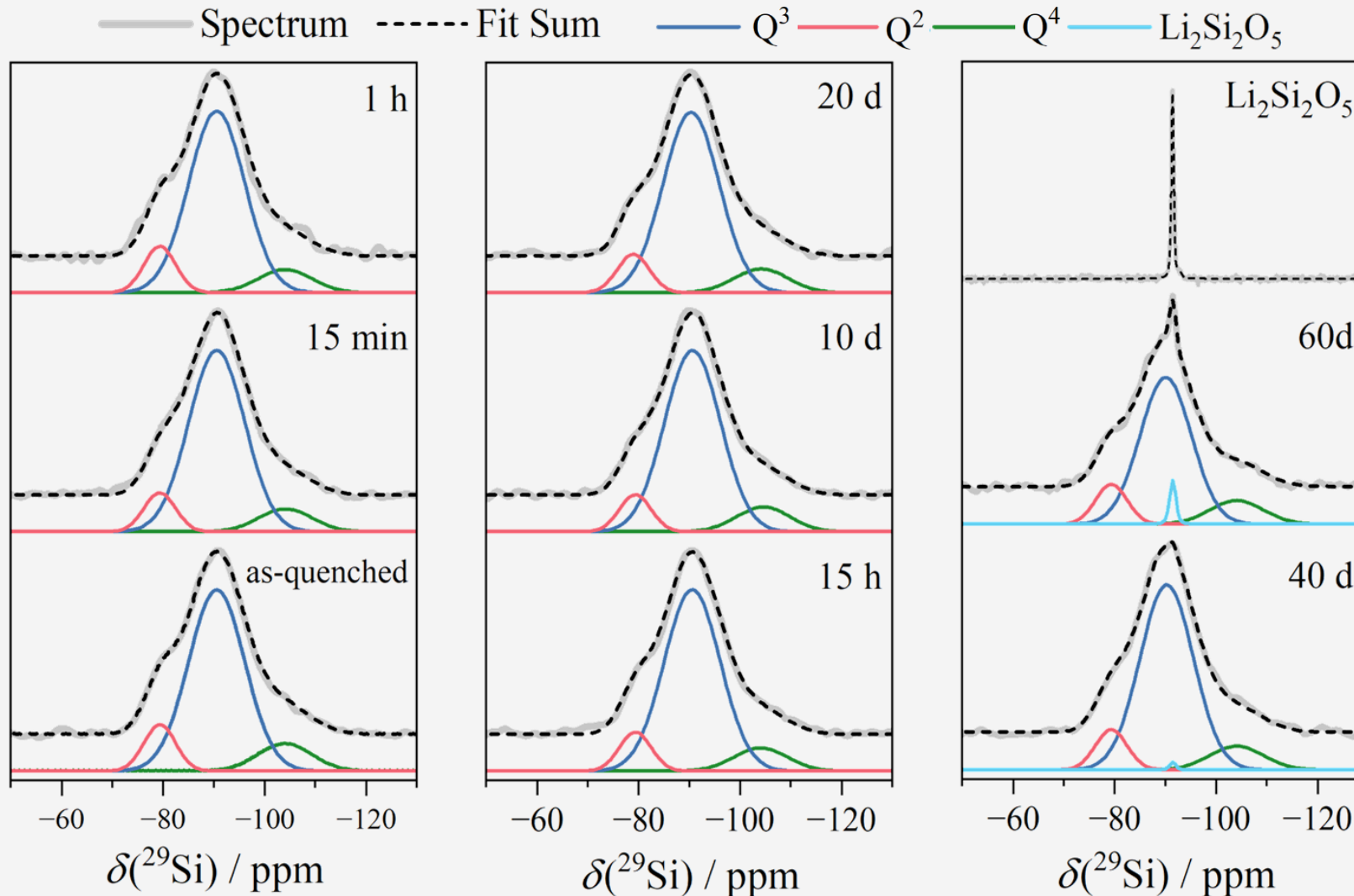


No detectable change in Q^n speciation upon annealing



No detectable change in Q^n speciation upon annealing

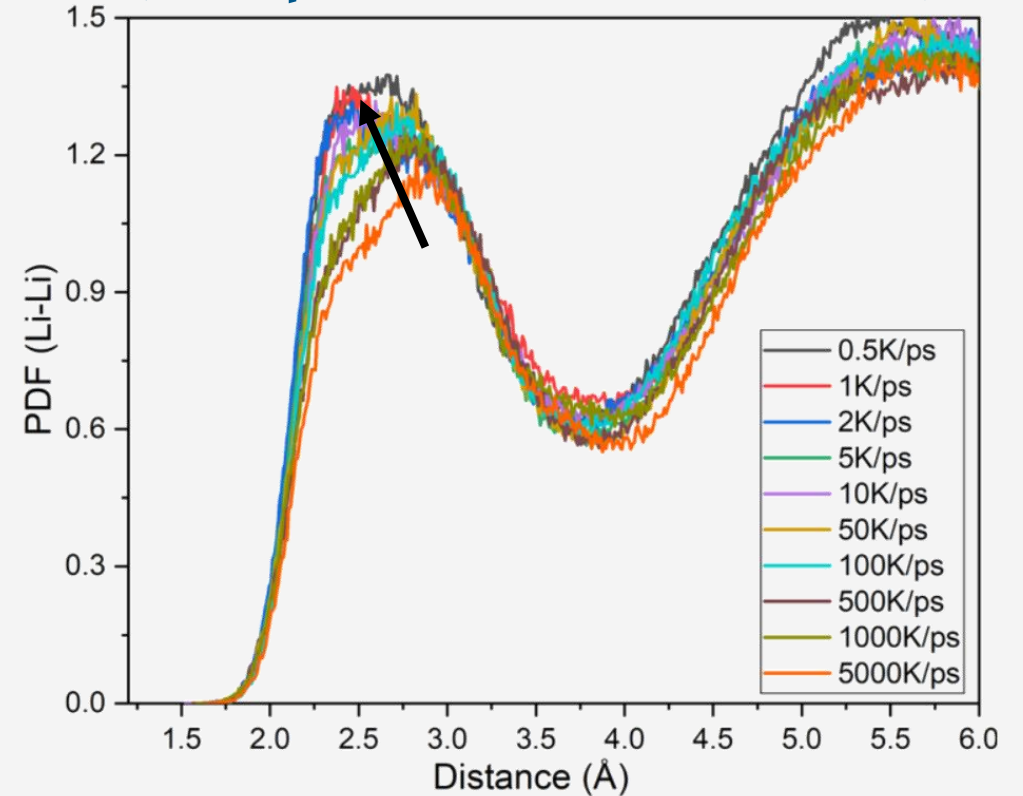
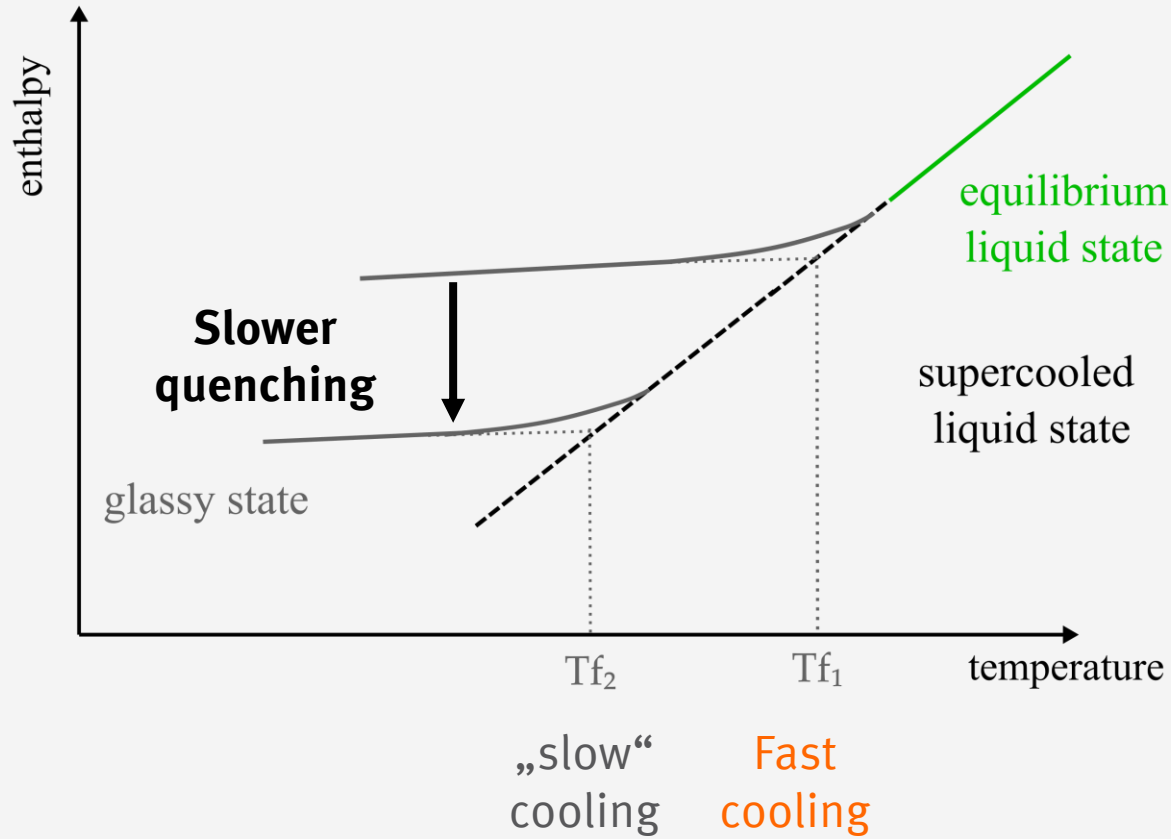
^{29}Si MAS NMR



Network former

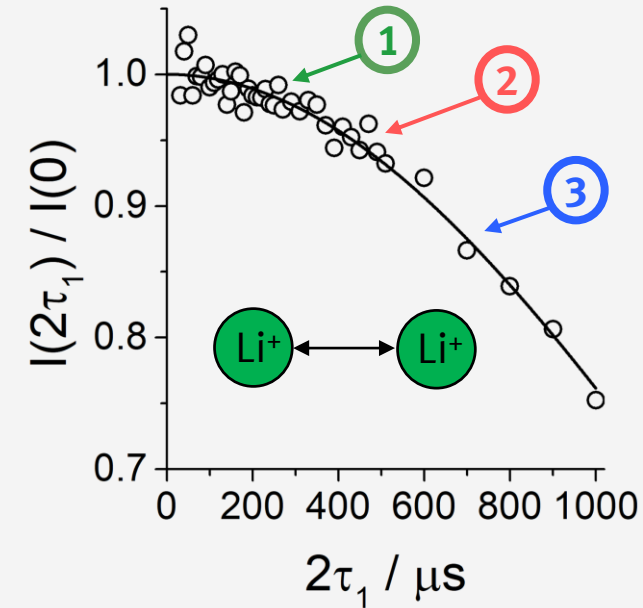
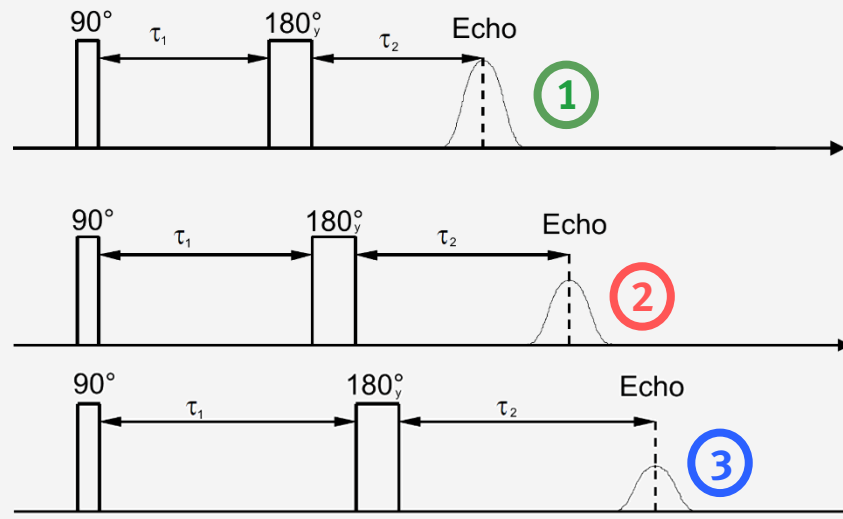
MD simulations indicate changes in network modifier distribution

MD simulations (Li-Li pair distribution function)



Network modifier

^7Li dipolar NMR can measure average Li–Li distances



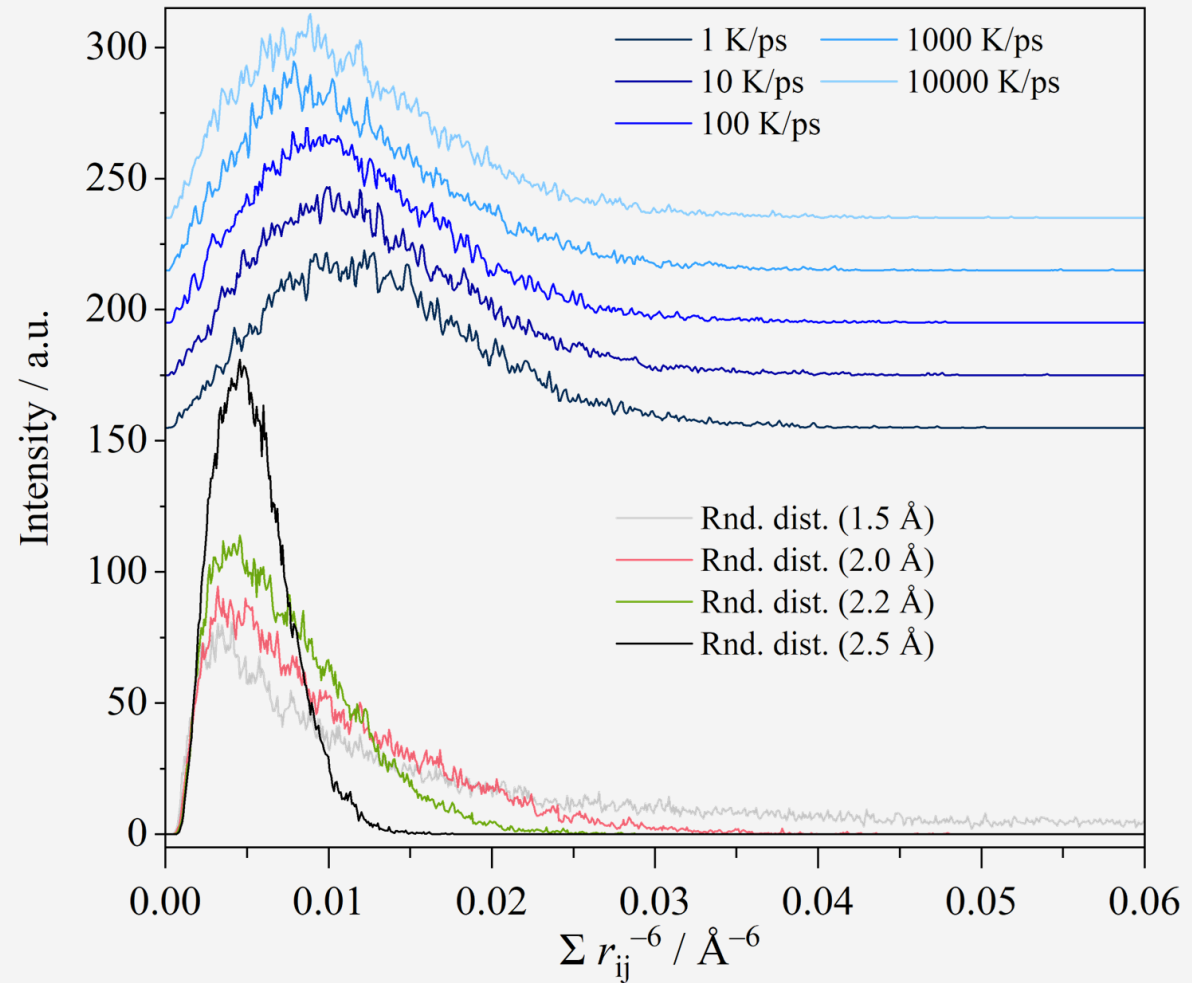
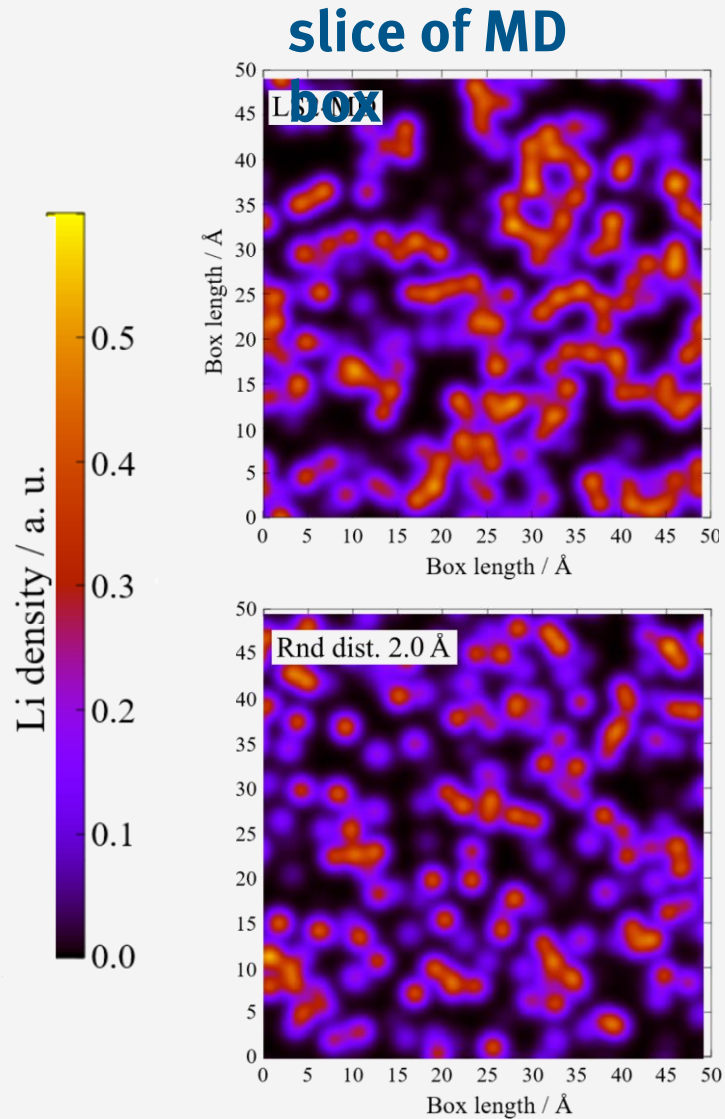
Experimental

$$\frac{I(2\tau)}{I(0)} = \exp \left[-\frac{M_{2E}}{2} (2\tau_1)^2 \right]$$

Theoretical

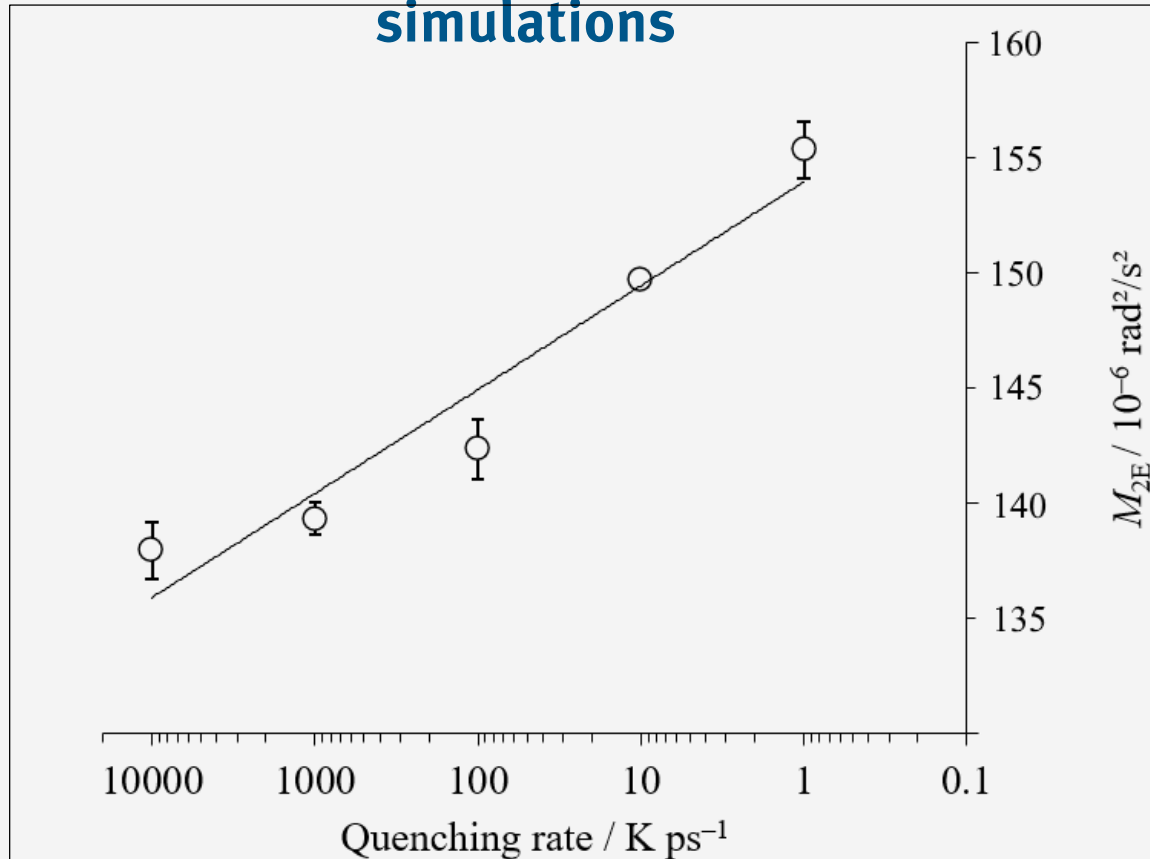
$$M_{2E} = 0.9562 \left(\frac{\mu_0}{4\pi} \right)^2 \gamma_{^7\text{Li}}^4 \hbar^2 \sum_{i \neq j} \frac{1}{r_{ij}^6}$$

MD simulations show some Li^+ clustering for decreasing T_f



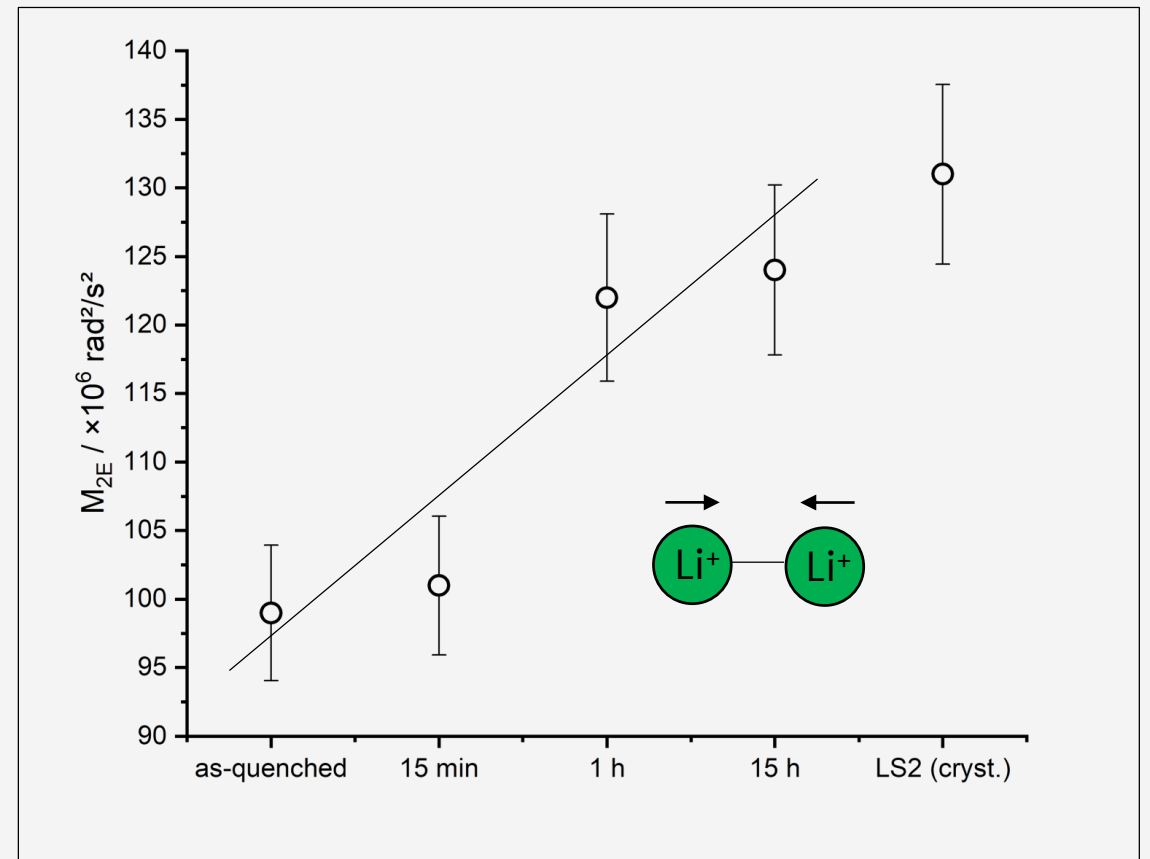
^7Li dipolar NMR confirms clustering with relaxation

MD
simulations



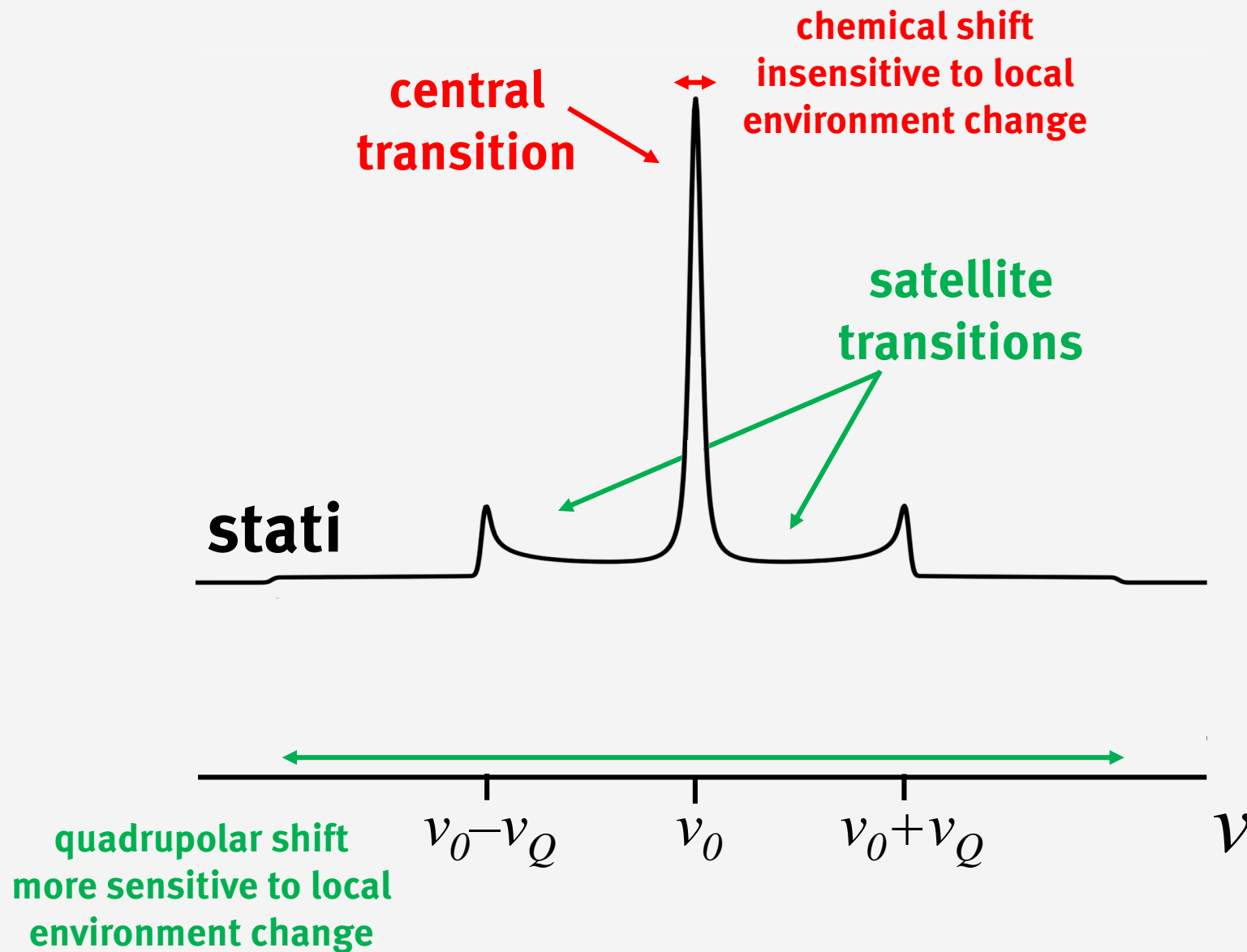
← fictive temperature

^7Li spin echo decay NMR

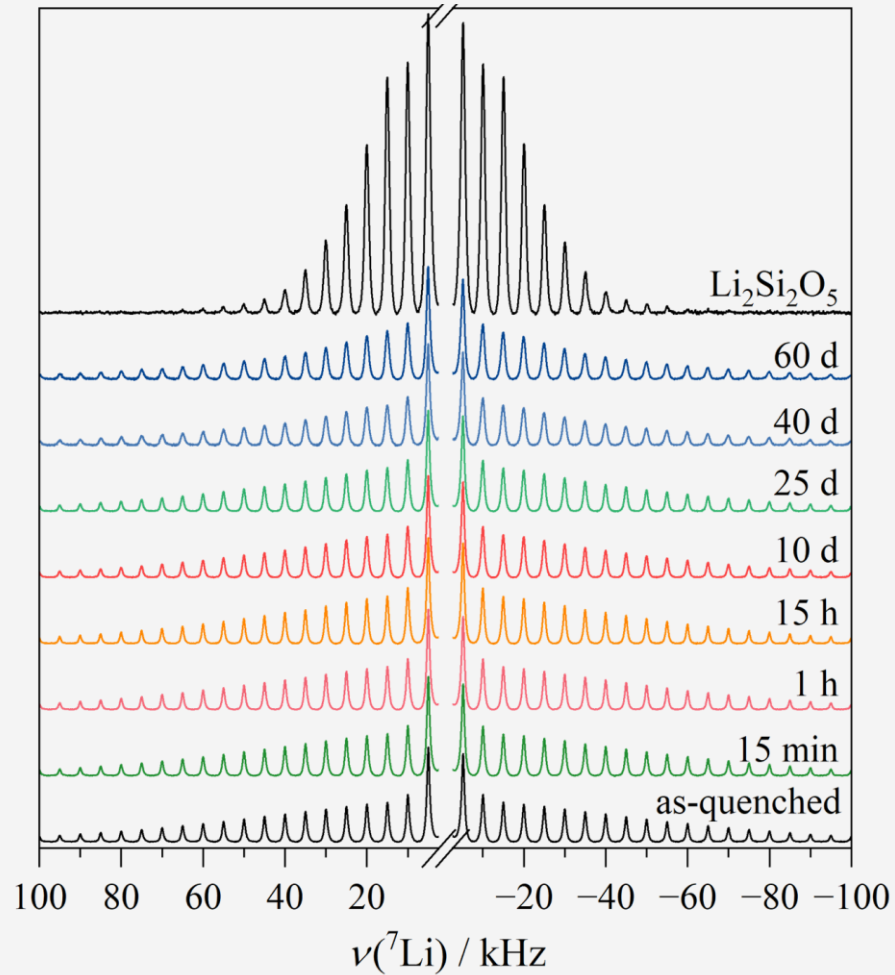


glass relaxation →

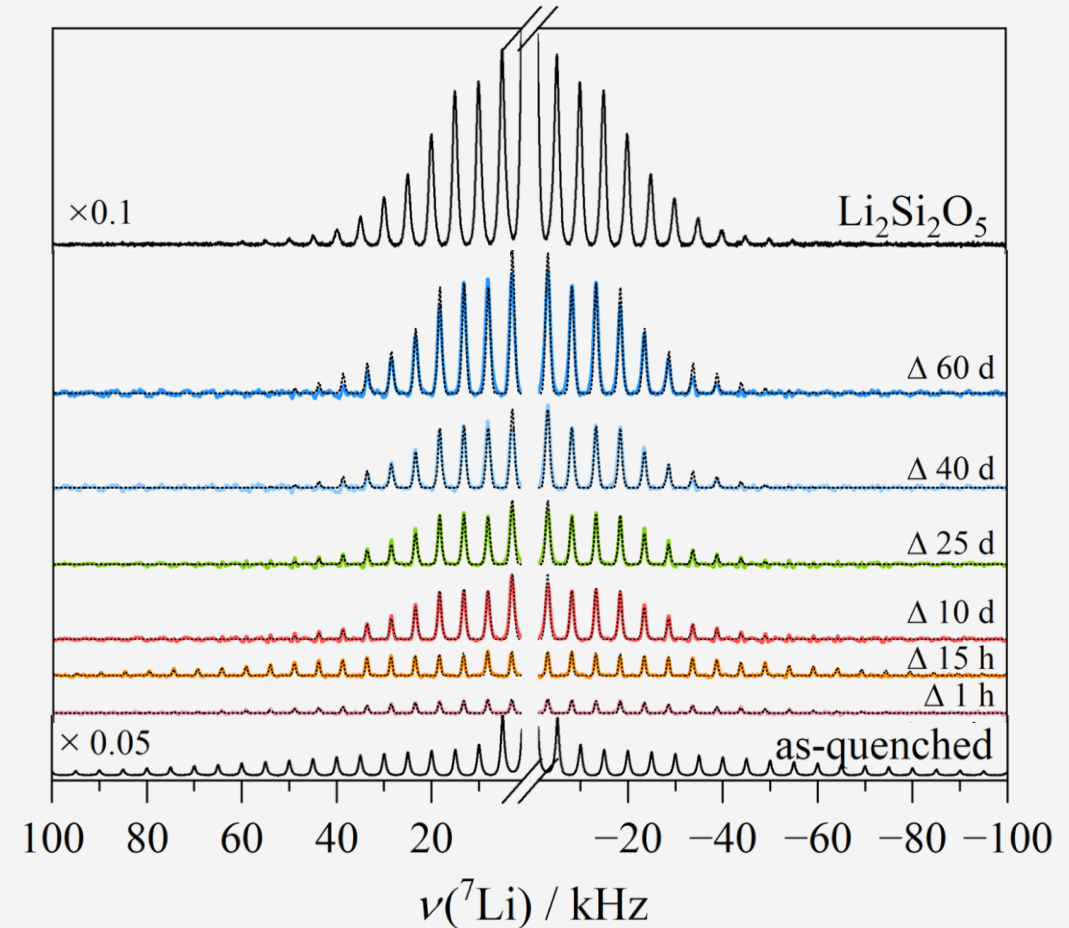
^7Li spin is also sensitive to changes in its local environment



^7Li difference spectroscopy exposes glass nucleation



SATRAS NMR



**Difference
spectra**

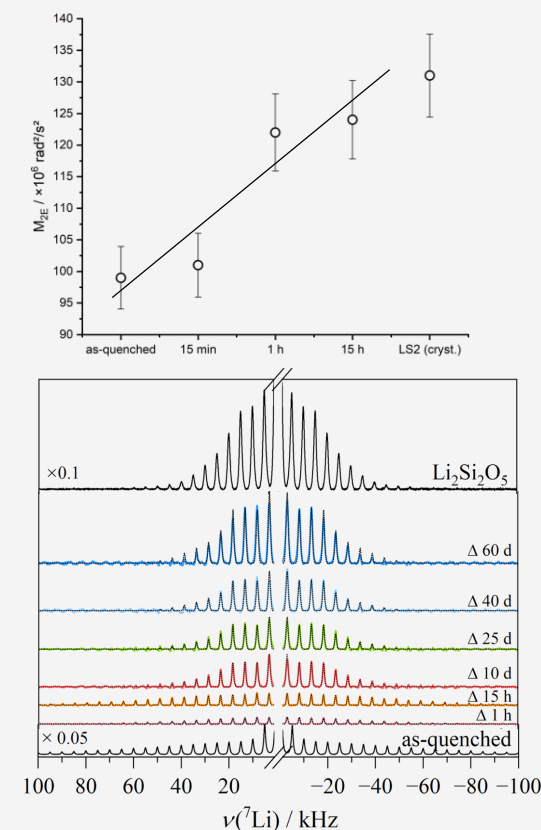
Network modifier

Sub T_g annealing of LS2 glass

Relaxation: reorganization of spatial Lithium ion distribution towards more clustering
 ^7Li spin echo decay spectroscopy

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Nucleation: profound change in electric field gradient distribution
 ^7Li SATRAS difference spectroscopy



**Most relevant: Changes in electrostatic interactions
network modifier \leftrightarrow NBOs.
 Q^n redistribution effects minor**

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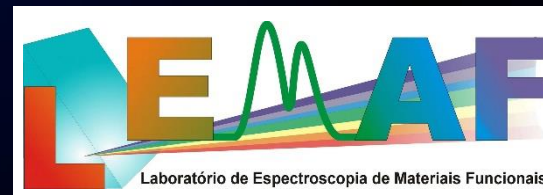
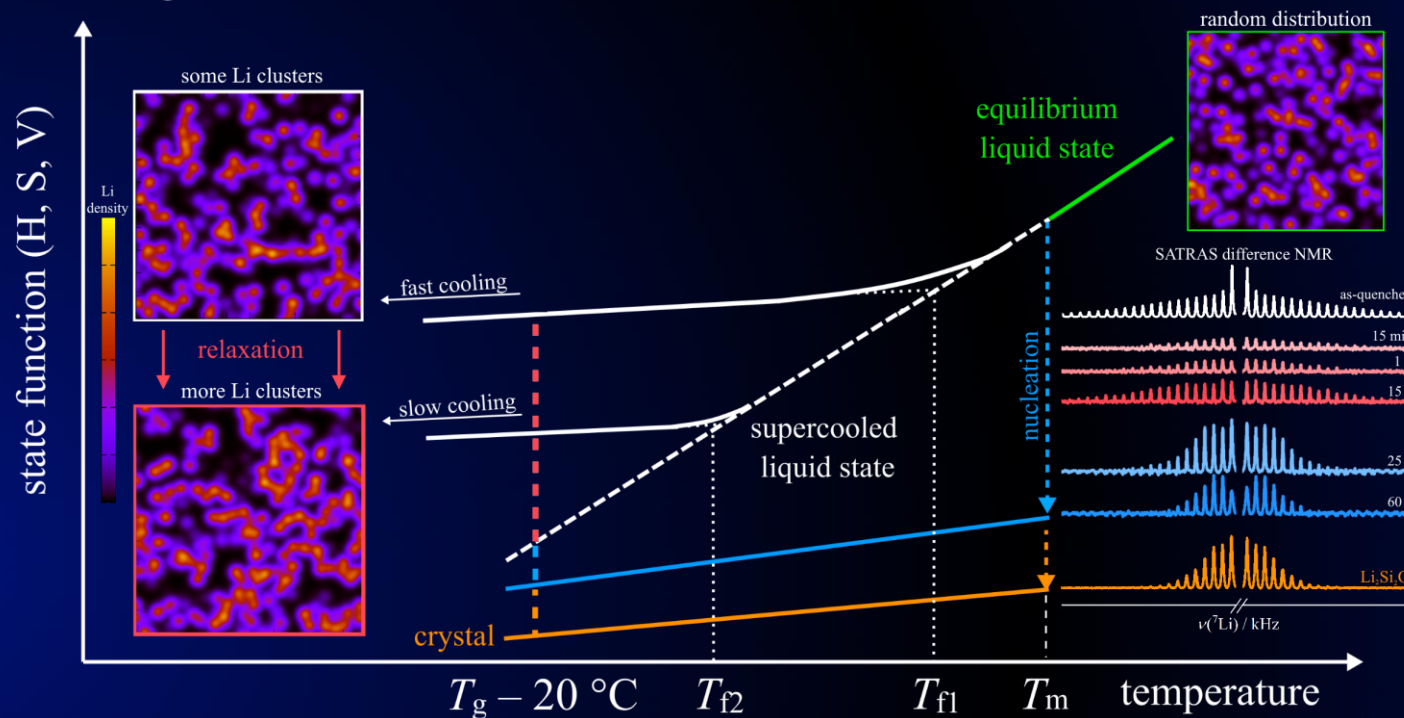
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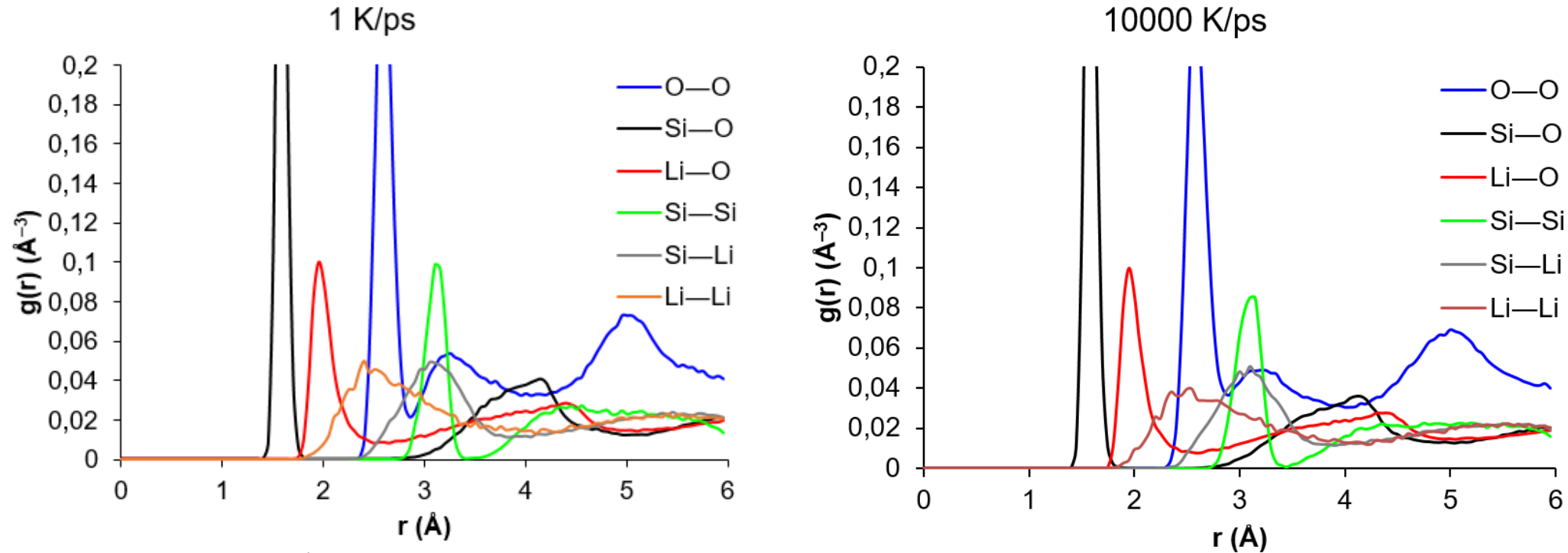
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Backup Slides

MD Simulations: effect of quenching rate

$$U(r) = \frac{z_i z_j e^2}{r} + D_{ij} \left\{ \left(1 - e^{-a_{ij}(r-r_0)} \right)^2 - 1 \right\} + \frac{C_{ij}}{r^{12}}$$



Pair	D_{ij} (eV)	a_{ij} (Å ⁻²)	r_0 (Å)	C_{ij} (eV Å ¹²)
Li ^{0.6} —O ^{-1.2}	0.001114	3.429506	2.681360	1.0
Si ^{2.4} —O ^{-1.2}	0.340554	2.006700	2.100000	1.0
O ^{-1.2} —O ^{-1.2}	0.042395	1.379316	3.618701	22.0